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chapter 8 chemistry test. STUDY. PLAY. atoms held together by sharing electrons. covalent bond. where are many elements found in nature exist as. molecules. neutral group of atoms joined together by covalent bonds. molecule. compounds that are formed when two or more atoms combine to form molecules are called.

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A)8, 2, and 3 B)7, 1, and 2 C)2, 7, and 4 D)2, 4, and 6 E)1, 5, and 7 6) Lattice energy is _____. A)the sum of electron affinities of the components in an ionic solid B)the energy required to convert a mole of ionic solid into its constituent ions in the gas phase C)the sum of ionization energies of the components in an ionic solid

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8.8 Chapter Summary. To ensure that you understand the material in this chapter, you should review the meanings of the bold terms in the following summary and ask yourself how they relate to the topics in the chapter. Any hydrocarbon containing either a double or triple bond is an unsaturated hydrocarbon. Alkenes have a carbon-to-carbon double bond.

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